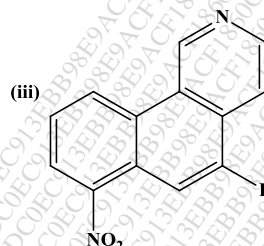
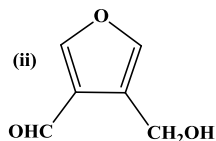
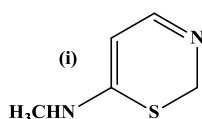
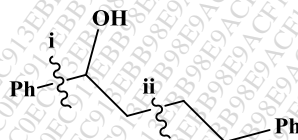
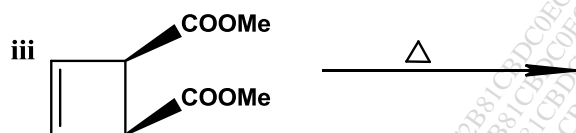
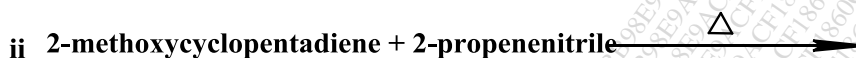


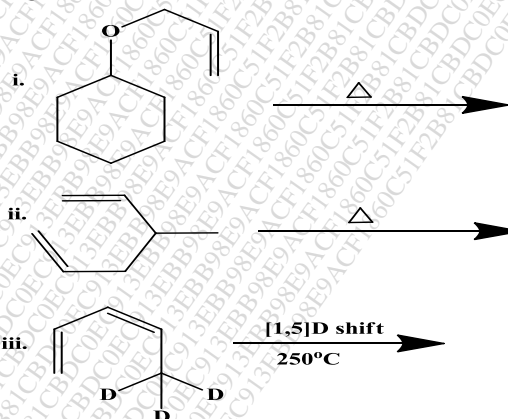
**Q. P. Code: 20760****N.B.: 1. All Questions are compulsory****Time: 3 Hours****Total Marks: 70****2. Figures to right indicate full marks****Q. 1. A. Give the IUPAC nomenclature of the following compounds****(03)****B. Define following terms: (i) Disrotatory motion (ii) E-factor (iii) Steroids** **(03)****C. Draw structures of the following: (i). Testosterone (ii) 5 $\beta$ -cholestan-3 $\beta$ , 6 $\beta$ -diol (in chair form)****(iii) 5 $\alpha$ -estrane chair form.** **(03)****D. Give suitable explanation for the statement 'Pyridine is a weaker base than aliphatic tertiary amines'.** **(02)****E. Depict the two disconnection strategies** **(02)****F. Depict hydrotalcite catalyzed condensation reactions.** **(02)****Q. 2. A. Write mechanism for (Any Two): (i). Friedlander synthesis (ii). Bischler-Napieralski****(iii). Paal Knorr synthesis for furan** **(04)****B. 'Thermal (4+2) cycloaddition of 1,3-butadiene with propenal is symmetry allowed', justify the statement using molecular orbitals** **(04)****C. Discuss the green strategies for MPV reduction.** **(03)****Q. 3. A. Depict mechanism for the following conversions (Any Two)** **(04)****(i). 2 moles of Ethylacetoacetate to 2,4,6-trimethylpyridine (ii). Benzil to 2,4,5-triphenylimidazole****(iii). Diethylmalonate to 2,4,6-trihydroxypyrimidine****B. Design retrosynthetic scheme and synthesis for Ibuprofen OR Benzocaine** **(04)****C. Compare merits and demerits of classical and zeolite catalyzed Friedel-craft acylation with an example.** **(03)****Q. 4. A. Write structures of products formed for the following reactions (Any Eight):** **(08)****i. Thiophene  $\xrightarrow{\text{conc. HNO}_3}$** **vi. 8-bromoisquinoline  $\xrightarrow{\text{NaNH}_2, \Delta}$** **ii. Furan + Maleic anhydride  $\xrightarrow{\Delta}$** **vii. Indole  $\xrightarrow{\text{HCN, HCl}}$** **iii. Pyrrole  $\xrightarrow{\text{EtOK, CHCl}_3}$** **viii. Imidazole  $\xrightarrow{\text{oleum, 100}^\circ\text{C}}$** **iv. Pyridine  $\xrightarrow{\text{peracetic acid}}$** **ix. Pyrimidine  $\xrightarrow{\text{PhMgBr, Ether}}$** **v. 6-Hydroxymethylquinoline  $\xrightarrow{\text{Vapour phase bromination}}$**

**Q. P. Code: 20760****B. Write structures of products formed for the following reactions:****(03)****Q. 5. A. Give suitable justification for the following statements:****(08)**

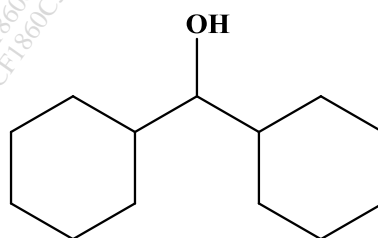
- Hoffmann degradation of  $3\alpha$ -trimethylammonium- $5\alpha$ -cholestanes gives  $5\alpha$ -cholestan-2-ene while  $3\beta$ -trimethylammonium- $5\beta$ -cholestanes do not form any product.
- Cholestan- $3\beta,5\alpha,6\beta$ -triol on reaction with ethylchloroformate in presence of pyridine gives  $3\beta$ -monocathylate while Cholestan- $3\beta,5\alpha,6\alpha$ -triol forms  $3\beta,6\alpha$ -dicathylated product.
- Indole undergoes Electrophillic Aromatic Substitution reaction preferably at which position and why?
- Furan, thiophene and pyrrole are all heteroaromatic compounds.

**B. Write reagents and reaction conditions for the following conversions:****(03)**

- Furan to 2-nitrofuran
- Isoquinoline to phthalimide
- 4-bromopyridine to 3,4-pyridyne

**Q. 6. A. Complete the following reactions with mechanisms (Any Two):****(04)****B. Draw resonating structures of the following:****(04)**

- Thiophene
- Quinoline
- Imidazole
- Pyridine

**C. Give retrosynthetic and synthetic pathway for the following target molecule:****(03)**

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